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## Summary:

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Textbook: Elementary Principles of Chemical Processes (3rd ... These resources were developed with support by the National Science Foundation under grant numbers (NSF DUE 0920640) and (NSF DUE 1322300). Any opinions, findings, and conclusions or recommendations expressed in this material are those of the author(s) and do not necessarily reflect the views of the National Science Foundation. Chapter 8 Balances on Nonreactive Processes Date: Table B.1 of Felder & Rousseau:  $\Delta H_{\text{H}_2\text{O}} = \hat{H}_{\text{H}_2\text{O}} - \hat{H}_{\text{H}_2\text{O}}^{\text{ref}}$  = The enthalpy of the water vapor entering the process will be calculated by the addition of the following changes in enthalpy: -Change of enthalpy of the liquid water from T ref to the boiling point temperature T bp. Felder and Rousseau Elementary Principles of Chemical ... Felder and Rousseau Elementary Principles of Chemical Processes 3.pdf. ... or download with email. Felder and Rousseau Elementary Principles of Chemical Processes 3.pdf. Download. Felder and Rousseau Elementary Principles of Chemical Processes 3.pdf.

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